$(\eta^{5}-C_{5}H_{5})Fe(CO)_{2}{P[=C(SiMe_{3})_{2}]_{2}}, the First Three-co-ordinate Metallo–Bis(methylene)phosphorane$

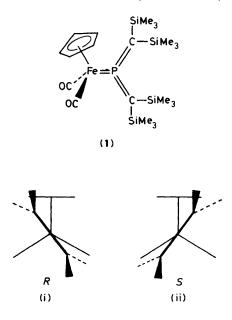
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The reaction of $CIP[=C(SiMe_3)_2]$ with $[(\eta^5-C_5H_5)Fe(CO)_2]K$ affords $(\eta^5-C_5H_5)Fe(CO)_2\{P[=C(SiMe_3)_2]_2\}$, the first example of a three-co-ordinate metallo-bis(methylene)phosphorane (metallophospha-allene).

Several examples of λ^5 -metallophosphoranes of the type R_4PML_n have been reported.¹ However, to the best of our knowledge metallophosphoranes with lower co-ordination numbers have not been prepared. We report the preparation and characterisation of the first example of a three-co-ordinate metallophosphorane (metallophospha-allene).

A mixture of $[(\eta^5-C_5H_5)Fe(CO)_2]K$ (2.13 mmol) and ClP[=C(SiMe_3)_2]_2^2 (2.09 mmol) in tetrahydrofuran (50 ml) was refluxed for 12 h. Removal of the solvent, followed by vacuum sublimation of the residue (40 °C, 0.1 Torr), afforded

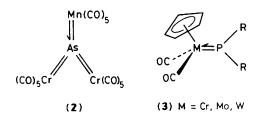


orange, crystalline (1) in 83% yield [m.p. 78 °C (decomp.)]. Unfortunately, crystals of (1) were unsuitable for X-ray diffraction study because of twinning problems. However, the composition $C_{21}H_{41}FeOPSi_4$ was established by high-resolution mass spectroscopy (calcd. for M^+ , 524.7258; found, m/z524.7251). In addition to the parent ion (m/z 524), prominent peaks were detected in the medium-resolution mass spectrum at m/z 436 and 384 which are due to the loss of Me₄Si and (C_5H_5)Fe(CO)₂, respectively.

Previous experimental work by Appel *et al.*³ has demonstrated that the (Me₃Si)₂C groups of XP[=C(SiMe₃)₂]₂ molecules are twisted out of the XPC₂ plane. *Ab initio* M.O. calculations on model three-co-ordinate phosphoranes⁴ suggest that this twisting is due to steric, and to a lesser extent, electronic effects. Furthermore, the vacant $3p_z$ orbital on phosphorus twists in response to rotation of the *P*-methylene units thereby resulting in an overall propeller-like arrangement. Such a conformation has, in fact, been observed⁵ for the anion {(fluoren-9-yl)P[=C(SiMe_3)₂]₂}⁻, which exists as *R*- and *S*-stereoisomers. As shown [(i) and (ii)], a similar situation exists for (1).

N.m.r. data[†] are in accord with the existence of *R*- and *S*-stereoisomers. Thus four Me₃Si groups are apparent in the ¹³C-{¹H} spectrum and the ¹H spectrum is somewhat complex in the Me₃Si region. Note also that the ³¹P-{¹H} n.m.r. chemical shift of (1) is 220 p.p.m. *upfield* of that of ClP[=C(SiMe₃)₂]₂. We attribute this observation to a donor-

[†] N.m.r. spectra, values in p.p.m., relative to SiMe₄ (¹H and ¹³C) and 85% H₃PO₄ external (³¹P) in [²H₆]benzene: ¹H (300 MHz) δ 4.08 (5H, d, J_{P-H} 48.0 Hz, C₅H₅) and 0.25 (36H, m, SiMe); ¹³C-{¹H} (75.5 MHz), δ 124.0 (d, J_{P-C} 26.3 Hz, C₅H₅), 69.8 (d, J_{P-C} 83.0 Hz, P=C), 2.52, 2.48, 2.43, and 2.38 (s, SiMe); ³¹P-{¹H} (32.2 MHz) δ -81.6(s).

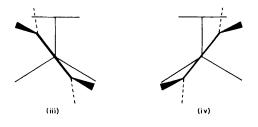


acceptor interaction between the a'' HOMO of the $C_5H_5Fe(CO)_2$ fragment⁶ and the vacant phosphorus $3p_z$ orbital.

Further support for the foregoing arguments is provided by the i.r. spectrum of (1) (toluene solution) which exhibits two CO stretching frequencies (1979 and 2040 cm⁻¹).[‡] Moreover, use of these vibrational spectroscopic data leads to a computed inter-carbonyl angle⁷ of $78 \pm 2^{\circ}$. This angle is *ca*. 95° when phosphorus is singly bonded to an (η^5 -C₅H₅)M(CO)₂ fragment. More acute angles are associated with the presence of phosphorus-metal multiple bonding.⁸

Finally, we draw attention to the fact that (1) is related to $(2)^9$ on the basis of the isolobal principle.¹⁰ Moreover, (1) is

[‡] In the solid state (Nujol mull) low-intensity shoulders are detected at 1963 and 2060 cm⁻¹. Tentatively this observation is attributed to the existence of the two high energy conformers (iii) and (iv). Note that in these conformers the vacant phosphorus $3p_z$ orbital is orthogonal to the a" donor orbital of the (C₅H₅)Fe(CO)₂ group.



isoelectronic with three-electron donor phosphido complexes (3).⁶ Both (2) and (3) have been shown to feature metal-group 15 multiple bonding.§

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§ Compound (1) also bears a resemblance to cationic iron alkylidene complexes. See M. D. Johnson in 'Comprehensive Organometallic Chemistry,' eds. G. Wilkinson, F. G. A. Stone, and E. W. Abel, Pergamon, Oxford, 1982, vol. 4, p. 331.